Title (3)	:	Dry Desulphurisation of Gas Streams Using KCC-1 Mesoporous Silica Functionalised with Deep Eutectic Solvents
Journal	:	Physical Chemistry Chemical Physics
Document Type	:	Article
Publisher	:	Royal Society of Chemistry 2024
UniKL Author	:	Raja Nazrul Hakim Raja Nazri
Link to Full Text	:	https://pubs.rsc.org/en/Content/ArticleLanding/2024/CP/D4CP03248A
Link to UniKL IR	:	
Link to Scopus Preview	:	https://www.scopus.com/inward/record.uri?eid=2-s2.0- 85209115177&doi=10.1039%2fd4cp03248a&partnerID=40&md5=c86824 ae71f2e84fc8f06f5dd595b24f
Abstract	:	Sulphur dioxide, a toxic gas pollutant, is mainly generated by the combustion of fossil fuels and the smelting of sulphur-bearing mineral ores. Removal of SO2 gas or desulphurisation can be accomplished in industries using a variety of processes; the most efficient is wet flue gas desulphurisation (FGD). However, wet FGD has challenges, such as the requirement for wastewater treatment, excessive water usage, and the necessity for chloride protective coating. Despite having a lesser adsorption capacity than wet FGD, dry FGD can efficiently remove SO2 from the effluent gas stream and avoid the issues associated with wet FGD, provided that the sorbents are modified and regenerable. An alternative dry desulphurisation strategy by using fibrous mesoporous silica (KCC-1) modified with deep eutectic solvents (DES), choline chloride–glycerol (DES1) and choline chloride–ethylene glycol (DES2) is studied in this paper. KCC-1 modified with DES1 is found to increase SO2 adsorption capacity to 4.83 mg g-1, which is 1.73 times greater than unmodified KCC-1 and twice higher than KCC-1 modified with DES2 attributed to the sorbent's high porosity. Increasing reaction temperature and SO2 concentration reduce the adsorption capacity to 1.73 mg g-1 and 2.73 mg g-1, respectively.