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Abstract	:	The need for accurate analytical glucose measurement techniques that are low-cost, highly sensitive, and user-friendly is crucial for glucose monitoring in healthcare and optimizing industrial processes. A simple and low-cost glucose monitoring was designed by synthesizing Zinc Oxide (ZnO) with Iron Oxide (Fe₂O₃) through a hydrothermal method by loading the composite onto reduced Graphene Oxide. The ZnO nanoparticles had a flower-like shape, while Fe₂O₃ had a sphere-like shape. These nanoparticles were evenly distributed on the wrinkled sheet surface of rGO with an average size of particles of ± 14.88 nm as observed via Transmission Electron Microscopy (TEM) and Field Emission Scanning Electron Microscopy (FESEM). The X-ray Diffraction (XRD) analysis exhibits overlapping peaks of diffraction of Fe₂O₃ and rGO at 24°, resulting from the reduction of GO. Cyclic Voltammetric analysis (CV) was used to determine the prepared composite's electrochemical properties, which was then further manufactured as a sensing probe. CV analysis revealed that the modified electrode's current response was higher than the bare electrode (ZnO and Fe₂O₃) with and without glucose. The modified glucose sensor shows a sensitivity of 504.23 $\mu\text{A mM}^{-1} \text{cm}^{-2}$ and a limit detection of glucose of 6.28 mM. Keywords: Electrochemical; reduced graphene oxide; metal oxide; glucose oxidation; ZnO/Fe₂O₃.